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Microfluidics: introduction

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Guell Park, Barcelone

Scaling laws

Quantity	Scaling
Intermolecular forces (Van der Waals)	l-7
Time	1-0
Capillary force	11
Flow velocity	1^{1}
Gravity force	13

The predominance of forces is different at smaller scales

requires developping fully new plumbing.

Adhesive force : van der Waals attraction



Une sétule peut supporter au maximum une force de 200 μ N tout en restant collée à une surface verticale. Ainsi théoriquement, si toutes les sétules avaient une force d'adhésion de 200 μ N, la force d'adhésion totale (6,5 millions de sétules) du gecko serait de 1300 N. Le gecko serait donc capable de supporter une masse d'environ 130 kg sur son dos tout en restant collé au plafond.















https://home.iitm.ac.in/arunn/laminar-flow-reversibility-why-does-the-blob-rewind.html

Fabrication of particles using laminar flows



Dandukuri et al., Nature Materials, 2006

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Fabrication of particles using laminar flows



Pregibon et al, Science 315, 2007

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Mean velocity 100 times smaller ! Cross section 100 smaller => Flow rate 10 000 times smaller

The hydrodynamic resistance



The hydrodynamic resistance

shape		$R_{ m hyd}$ expression	R_{hyd} [10 ¹¹ $\frac{P_{a.s}}{m^3}$]	shape	. <u></u>	$R_{\rm hyd}$ expression	$\frac{R_{\rm hyd}}{[10^{11} \ \frac{\rm Pas}{\rm m^3}]}$
circle	a	$\frac{8}{\pi} \eta L \frac{1}{a^4}$	0.25	rectangle	h	$\frac{12\eta L}{1-0.63(h/w)}\frac{1}{h^3w}$	0.51
ellipse	b a	$\frac{4}{\pi} \eta L \frac{1 + (b/a)^2}{(b/a)^3} \frac{1}{a^4}$	3.93	square	$h \begin{bmatrix} h \\ h \end{bmatrix} h$	$28.4 \eta L \frac{1}{h^4}$	2.84
triangle	$a \sqrt{a}$	$\frac{320}{\sqrt{3}} \eta L \frac{1}{a^4}$	18.5	parabola	hw	$\frac{105}{4} \eta L \frac{1}{h^3 w}$	0.88
two plates	h	$12 \eta L \frac{1}{h^3 w}$	0.40	arbitrary	PA	$\approx 2 \eta L \frac{P^2}{A^3}$	-





Guillot et al., Langmuir, 2006.



Flows are not always simple laminar flows

pathogen Pseudomonas aeruginosa



MIXING

Biofilm streamers that initiate on corners rapidly expand. Wall-attached biofilm is a necessary precondition for streamer formation

Drescher et al., PNAS 2012



At low Reynolds numbers



Negligible diffusion

No turbulence

Nearly two dimensional flow

NO MIXING!

Diffusion with « hands »



Generating concentration gradients



Dertinger et al., Anal. Chem. 2001, 73, 1240-1246

Generating concentration gradients



How to mix at low Reynolds numbers?

Convection-diffusion equation

• Convection equation for an incompressible liquid :

$$\frac{\partial C}{\partial t} + \vec{u}\cdot\vec{\nabla}C = D\Delta C + q$$



Mixing in a non stationary flow : Chaotic regimes

A regime is said chaotic if ,



And so on... (baker transformation)

In chaotic regimes, two close particles separate exponentially



• Glycérine, Re=1





From Ottino's book : « Chaotic Advection »

Passive chaotic micromixer

Corrugations produce a three-dimensional flow!



Stroock et al., Sciences 2004





Passive chaotic micromixer



Active chaotic micromixer



Droplet based microfluidics

Droplet-based microfluidics

- Applications :
 - droplet = unit system

- Microreactor

- Decrease of cross-pollution (appart liquid/liquid extraction)
- → Necessicity to controle elementary operation:

fabrication, fusion, breakup, transport, storage...

• Models : microporous (stone, lung..)

Definition: surface tension



 $\gamma\,$ Is the surface tension. Its is an energy per unit surface or a force per unit length.



Laplace law demonstration for a sphere





Laplace law : sphere



Gravity vs. Capillary forces

• Bond number

$$Bo = \frac{\text{gravity force}}{\text{capillary force}} = \frac{\rho l^3 g}{\gamma l} = \frac{\rho l^2 g}{\gamma}$$



• Capillary length



in situ fabrication

• Capillary number:





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hight => viscosity dominates

small => surface tension dominates





Fabrication of more complex systems

Encapsulation:





Janus droplet:



Using droplets for antibiograms





Using droplets for antibiograms



Baraban et al. Lab Chip 2011

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Wetting film and triple line



The spreading coefficient



S < 0 Partial wetting or desorption





Capillary imbibition: vertical tube – Jurin law



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Capillary imbibition: horizontal tube – Washburn

$$g = -ge_z$$

$$p^* \bullet \qquad h$$

$$p^* \bullet \qquad p^* - \Delta p_{surf} \bullet p^*$$

$$L(t)$$

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Driving force: $\Delta p_{surf} = \frac{2\gamma}{h} \cos \theta$ Viscous force: $\Delta p = (12\mu L/h^3 w)Q$

$$\frac{\mathrm{d}L(t)}{\mathrm{d}t} = V_0 = \frac{Q}{wh} \sim \frac{h^2 \Delta p_{surf}}{12\mu} \frac{1}{L(t)}$$
$$L(t) = h \sqrt{\frac{t}{\tau_{adv}}} \quad \text{with} \quad \tau_{adv} \equiv \frac{3\mu h}{\gamma \cos \theta}$$

Washburn, Phys. Rev. 1921

Electrowetting : triple line



1<u>00</u>µm



Electrowetting - EWOD











State of the art: the lubrication problem







State of the art

Pressure drop along the droplet in the lubrication film (meniscus + flat film): $\int \delta n^{sf} - 4.94 C a^{2/3} \frac{4\gamma}{\gamma}$

$$-\begin{cases} \delta p^{sli} = \left(6.82Ca^{2/3} + 0.94\frac{2l}{H}Ca^{1/3}\right)\frac{4\gamma}{H} \\ \delta p^{rol} = 6.22Ca^{2/3}\frac{4\gamma}{H} \end{cases}$$
 Pressure drop in the sheared lubrication film

Comparison, pressure drop in the stress-free case vs. Poiseuille flow

$$\begin{split} \delta p^{\rm Pois} &= 8\eta L_t U/r_t^2 \\ r_t &= 1 \mbox{ mm}, \ Ca = 10^{-3} \implies \delta p^{sf} \sim \delta p^{\rm Pois} \mbox{ for } L_t = 12 \mbox{ mm} \end{split}$$

Cantat, Phys. Fluids, 2013

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A kitchen experiment:



Vanapalli et al., Lab Chip, 2009





γ^+ Alcohol evaporation γ wine

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MARANGONI EFFECTS

glass



Droplet displacement

• Wetting film: a) Thermocapillarity

b) Electrocapillarity

c) Solutocapillarity



State of the art

Thermocapillarity: state of the art (wetting film)

Authors	Geometry	Scale	Direction
Young, Goldstein (1959)	3D	micron	Hot, low γ
Sammarco (1999) Lajeunesse (2003)	1D (tube)	mm	Hot, low γ
Brathukin and Zuev (1984)	2D (Hele-Shaw cell)	mm	Hot, low γ



Thermocapillarity vs. thermomechanical

Direction of motion



cold

hot



Involved machanisms



Towards cold

 $\frac{\partial e}{\partial x}$

Thermocapillarity vs. thermomechanical

Thermomechanical actuation: a versatile technology for droplet-based microfluidics

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de

 T^+

Anchor rails

Application : bubble trap



Droplet Fusion

Marie-Caroline Jullien, Microfluidics



Solutocapillary



Baroud et al., lab Chip 2007

Particle in an electrical field



b)





Cell 10 µm for two charges:

Positive dielectrophoresis

Droplet sorting

Dipolar moment m toward E



Negative dielectrophoresis

Dipolar moment m opposite to E





New materials

Baret et al., Lab Chip 2009

Droplet-based microfluidics

Morphological Comparison of PVA Scaffolds Obtained by

Gas Foaming and Microfluidic Foaming Techniques





Colsi et al., Langmuir, 2013

Design and characterization of bubble phononic crystals



Leroy et al., APL, 2009

Acoustic insulation









New paradigm: closed cell foams



Liquid foam